

Sesquiterpenoids from *Inula linearifolia* Inhibit Nitric Oxide ProductionLi-Yue Nie,[†] Jiang-Jiang Qin,[†] Ying Huang,[†] Lan Yan,[†] Yu-Bo Liu,[†] Yue-Xing Pan,[‡] Hui-Zi Jin,^{*,†} and Wei-Dong Zhang^{*,†,‡}

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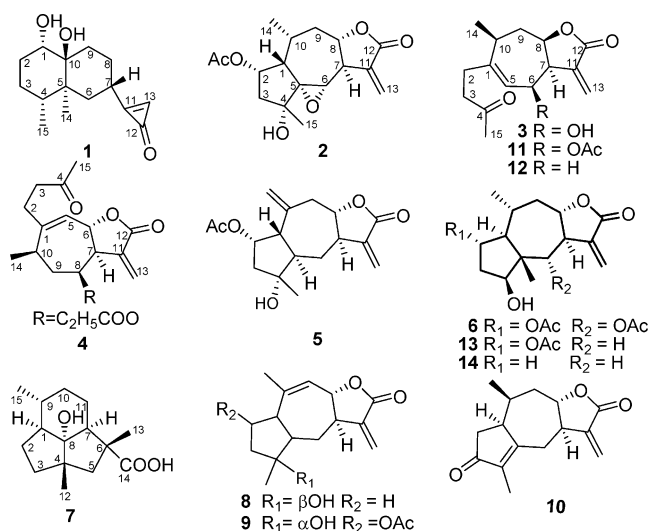
Fourteen sesquiterpenoids including six new analogues (**1–6**) were isolated from the aerial parts of *Inula linearifolia*. The structures of the new sesquiterpenoids were determined by spectroscopic data analysis, and the structures of **1** and **6** were confirmed by single-crystal X-ray diffraction data and the modified Mosher method. All 14 compounds were examined for the inhibition of LPS-induced nitric oxide production in RAW264.7 macrophages. Compounds **6**, **13**, and **14** were found to inhibit nitric oxide production potently, with IC₅₀ values of 0.11, 0.25, and 0.10 μM, respectively.

The genus *Inula* (Asteraceae family) is composed of about 100 plant species, a number of which are used as traditional herbal medicines throughout the world.¹ Recently, much attention has been paid to *Inula* species due to their diverse biological activities, which include anti-inflammatory, antitumor, and bactericidal effects.^{1,2} Plants of *Inula* species are rich sources of sesquiterpenes, such as eudesmanes, macrophylic acids, elemanes, germacranes, guaianes, bis-sesquiterpenes, and sesquiterpene-monoterpene dimers.¹ *Inula linearifolia* Turcz. is widely distributed in China and is used in the traditional Chinese medicine “JinFeiCao”. However, only one sesquiterpenoid, britanin, has previously been reported from this plant.³ As part of our ongoing search for biologically active sesquiterpenoids from *Inula* species, six new (**1–6**) and eight known sesquiterpenoids (**7–14**) were isolated from the aerial parts of *I. linearifolia*. Herein, we describe the isolation and structure elucidation of these six new sesquiterpenes and the inhibitory activities of all 14 compounds against lipopolysaccharide (LPS)-induced nitric oxide (NO) production in RAW264.7 macrophages.

eluants, to afford six new (**1–6**) and eight known (**7–14**) sesquiterpenoids.

Compound **1** had the molecular formula C₁₅H₂₂O₃ by positive HRESIMS at *m/z* 251.1651 [M + H]⁺, indicating five degrees of unsaturation. Its IR spectrum exhibited the presence of two OH groups (3450 and 3268 cm⁻¹) and a cyclopropane (1845, 1556 cm⁻¹) group. The ¹H, ¹³C, and DEPT NMR spectra showed the presence of one singlet and one methyl doublet [δ_{H} 0.92 (3H, s, H₃-14), δ_{C} 17.0; 0.77 (3H, d, *J* = 6.9 Hz, H₃-15), δ_{C} 15.5], five methylenes, four methines, including an oxygen-bearing carbon [δ_{H} 3.42 (m); δ_{C} 76.5 (C-1)], one olefinic carbon [δ_{H} 8.74 (1H, d, *J* = 12.5 Hz); δ_{C} 147.9 (C-13)], four quaternary carbons, including an olefinic carbon [δ_{C} 161.9 (C-11)], one carbonyl [δ_{C} 176.1 (C-12)], and one oxygenated quaternary carbon [δ_{C} 74.3 (C-10)] (Tables 1 and 2). Analysis of the ¹H–¹H COSY plot of **1** suggested the following correlations: H-1/H₂-2/H₂-3/H-4/H₃-15 and H₂-6/H-7/H₂-8/H₂-9 (Supporting Information). The connections of these moieties through a quaternary carbon was established by analyzing HMBC correlations. In the HMBC spectrum, the cross-peaks from H-1 to C-3, C-5, and C-10; H₂-9 to C-1, C-5, and C-10; H₃-14 to C-4, C-6, and C-10; H₃-15 to C-3 and C-5; H-7 to C-12 and C-13; and H-13 to C-12 indicated that **1** had a planar structure. The relative configuration of **1** was confirmed by single-crystal X-ray diffraction (Figure 1), which was consistent with the NOESY correlations: H₃-15 to H-3 α ; H₃-14 to H-3 α and H-6 α ; and H-1 to H-3 β and H-4. The absolute configuration of **1** was determined by a combination of the modified Mosher method^{4,5} and X-ray data (Figure 2). Thus, the absolute configuration of **1** was established as 1*S*, 4*R*, 5*S*, 7*S*, and 10*S*, and thus the structure of **1** was elucidated and it was named ineariifolianone.

Compound **2** was isolated as a light yellow gum. The molecular formula was established as C₁₇H₂₂O₆ by HRESIMS at *m/z* 321.1339 [M – H]⁻, indicating seven degrees of unsaturation. Its IR spectrum exhibited absorption bands corresponding to an OH group (3427 cm⁻¹), an ester carbonyl (1738 cm⁻¹), and an olefinic bond (1627, 1456 cm⁻¹). The similarities between the NMR spectra of **2** (Tables 1 and 2) and gaillardin (**9**) suggested that **2** was also a sesquiterpene lactone with a guaianolide skeleton, esterified with an acetyl group.⁶ The main difference between the ¹³C NMR spectra of **2** and **9** was that the signals of the two olefinic carbons [δ_{C} 136.2 (C-10), δ_{C} 128.2 (C-9)] in **9** shifted upfield to δ_{C} 34.1 and δ_{C} 43.9, whereas the methine group at δ_{C} 51.0 (C-5) and the methylene group at δ_{C} 34.4 (C-6) in **9** moved downfield to δ_{C} 74.3 and δ_{C} 56.3 in **2**, respectively. These observations suggested that the olefinic carbons in **9** were hydrogenated in **2** and that an epoxide group was present between C-5 and C-6 in **2** due to the remaining unsaturation. The assumption was confirmed by 2D NMR data, in which HMBC correlation of H-2 to the carbonyl carbon at δ_{C} 170.3 indicated that the acetoxy group was located at C-2. In NOESY spectra, the



Results and Discussion

The petroleum ether and CHCl₃-soluble fractions of the MeOH extract of the aerial parts of *I. linearifolia* were subjected to repeated column chromatography over silica gel, MCI, Sephadex LH-20, and preparative HPLC, using various solvent systems as

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Table 1. ^1H NMR Spectroscopic Data of **1–6** (J values in Hz) (500 MHz)

position	1 ^a	2 ^b	3 ^b	4 ^b	5 ^b	6 ^b
1	3.42 m	1.98 dd (12.5, 3.5)			2.51 dd (12.5, 5.0)	1.91 m
2	1.51 m; 2.20 m	5.02 m	2.59 m; 2.59 m	2.54 m; 2.54 m	5.30 ddd (11.5, 8.5, 6.0)	4.90 t (5.0)
3	1.59 m; 1.21 m	2.44 m; 1.81 m	2.29 m; 2.29 m	2.27 m; 2.27 m	2.20 m, 1.98 m	1.83 m, 2.07 m
4	1.90 m					4.13 dd (11.0, 9.0)
5			5.30 d (2.0)	5.08 d (2.5)	2.27 ddd (12.0, 9.5, 3.0)	
6	1.87 m; 2.08 m	3.23 d (5.5)	4.69 dd (10.0, 1.5)	5.69 dd (11.0, 5.0)	1.16 m; 2.38 m	5.05 d (12.0)
7	3.05 m	2.71 m	3.28 m	3.46 ddd (11.5, 8.5, 3.0)	2.71 m	3.04 m
8	2.27 m; 2.06 m	4.06 ddd (12.5, 9.5, 2.0)	4.61 ddd (12.5, 9.5, 3.5)	4.64 ddd (12.0, 8.5, 3.0)	3.91 dt (15.0, 3.0)	4.53 m
9	2.31 m; 1.24 m	2.44 m; 1.53 m	2.17 m; 1.80 m	2.10 m; 1.86 m	2.59 m; 3.06 dd (14.0, 3.0)	1.45 m; 2.45 m
10		1.81 m	2.36 m	2.39 m		1.90 m
13	8.74 d (12.5)	5.88 d (3.5)	6.36 d (2.0)	6.31 d (2.5)	6.20 d (3.0)	6.15 d (3.0)
		6.36 d (3.5)	6.09 d (2.0)	5.72 d (2.5)	5.53 d (3.0)	5.39 d (3.0)
14	0.92 s	1.07 d (6.5)	1.14 d (6.0)	1.13 d (7.0)	4.99 d (3.0); 5.15 d (3.0)	0.99 d (6.0)
15	0.77 d (6.9)	1.29 s	2.17 s	2.15 s	1.23 s	1.03 s
2-OAc		2.07 s			2.04 s	2.05 s
6-OAc						2.25 s
8-C ₂ H ₅ COO				2.39 m; 2.39 m	1.16 t (7.0)	

^a Recorded in methanol-*d*₄. ^b Recorded in CDCl₃.**Table 2.** ^{13}C NMR Spectroscopic Data of **1–6** (125 MHz)

position	1 ^a	2 ^b	3 ^b	4 ^b	5 ^b	6 ^b
1	76.5, CH	53.5, CH	141.9, C	142.7, C	51.0, CH	51.1, CH
2	29.8, CH ₂	75.3, CH	42.3, CH ₂	42.3, CH ₂	75.6, CH	75.3, CH
3	26.7, CH ₂	44.7, CH ₂	30.0, CH ₂	30.0, CH ₂	48.0, CH ₂	36.3, CH ₂
4	36.0, CH	70.9, C	207.7, C	207.7, C	78.5, C	72.9, CH
5	40.4, C	73.4, C	129.3, CH	125.9, CH	53.5, CH	51.1, C
6	36.9, CH ₂	56.3, CH	67.7, CH	70.0, CH	27.1, CH ₂	76.5, CH
7	34.3, CH	50.8, CH	48.9, CH	46.1, CH	49.3, CH	52.4, CH
8	23.2, CH ₂	78.0, CH	77.2, CH	76.9, CH	83.5, CH	76.1, CH
9	29.5, CH ₂	43.9, CH ₂	37.0, CH ₂	36.9, CH ₂	42.1, CH ₂	44.0, CH ₂
10	74.3, C	34.1, CH	35.2, CH	35.1, CH	141.8, C	30.2, CH
11	161.9, C	137.7, C	136.5, C	135.8, C	139.6, C	138.6, C
12	176.1, C	168.8, C	169.9, C	169.3, C	169.9, C	168.8, C
13	147.9, CH	121.9, CH ₂	125.4, CH ₂	125.0, CH ₂	119.1, CH ₂	120.0, CH ₂
14	17.0, CH ₃	19.9, CH ₃	20.7, CH ₃	20.6, CH ₃	116.5, CH ₂	15.9, CH ₃
15	15.5, CH ₃	24.9, CH ₃	29.9, CH ₃	29.9, CH ₃	24.6, CH ₃	20.4, CH ₃
2-OAc		170.3, C			170.4, C	170.2, C
		21.1, CH ₃			21.2, CH ₃	21.2, CH ₃
6-OAc						172.6, C
						21.2, CH ₃
8-C ₂ H ₅ COO				173.3, C		
				27.6, CH ₂		
				9.0, CH ₃		

^a Recorded in methanol-*d*₄. ^b Recorded in CDCl₃.

following correlations was observed: H-2 to H₃-15 and H-10; H-6 to H-8 and H₃-15; H-8 to H-6 and H-10. On the basis of these findings and by comparing its NMR data with that of the known compound gaillardin, **2** was identified as 5 α ,6 α -epoxy-2 α -acetoxy-4 α -hydroxy-1 β ,7 α -guaia-11(13)-en-12,8 α -olide.

Compound **3** was obtained as a colorless gum and exhibited a molecular ion $[\text{M} + \text{H}]^+$ at m/z 265.1445 in the HRESIMS, which corresponded to the molecular formula C₁₅H₂₀O₄. Except for a signal of an additional oxygenated C atom resonating at δ_{C} 67.7 (C-6), its ^1H and ^{13}C NMR spectra resembled those of tomentosin (**12**).⁷ These observations together with a 16 Da mass surplus indicated that **3** was a hydroxylated derivative of tomentosin. The relative configuration of **3** was determined using key NOESY correlations, namely, H-6 to H-9 α , H-9 α to H-10, H-6 to H-5, H-5 to H-3 α , and H-3 α to H-10. On the basis of the above-mentioned observations and literature values,⁷ **3** was identified as 6 β -hydroxytomentosin.

HRESIMS data showed **4** had the molecular formula C₁₈H₂₄O₅. The ^1H and ^{13}C NMR data closely resembled those of inusoniolide,⁸ but three additional carbon resonances were observed at δ_{C} 9.0, 27.6, and 173.3 in **4**, which were assigned to a propionyl moiety using HMQC and HMBC data. These observations, coupled with HMQC and HMBC analyses, revealed that **4** was a C-8-propionylated derivative of inusoniolide. Thus, the structure of **4** was determined as 8 β -propionylinusiolide.

Compound **5** was obtained as light yellow oil. HRESIMS gave a quasimolecular ion peak $[\text{M} - \text{H}]^-$ at m/z 305.1402 consistent with the molecular formula C₁₇H₂₂O₅. The ^1H and ^{13}C NMR data of **5** resembled those of the reported otoimbricatin B,⁹ suggesting that they have the same planar structure, which was also consistent with the 2D NMR data. Furthermore, the relative configuration of **5** was identical to that of **2** on the basis of NOESY correlations of H-1/H-2/H-8, H-2/H₃-15, and H-5/H-6 α /H-7 (Figure 3). Thus, **5** was identified as 2 α -acetoxy-4 α -hydroxy-1 β -guaia-11(13),10(14)-dien-12, 8 α -olide.

Compound **6** was obtained as colorless, tetragonal crystals after crystallization from MeOH. HRESIMS established a molecular formula of C₁₉H₂₆O₇ ($[\text{M} + \text{H}]^+$ at m/z 367.1767). A literature search revealed that the spectroscopic data of **6** resembled those of the known pseudoguaiaonlide-type sesquiterpene britanin.³ X-ray diffraction analysis of **6** (Figure 1) established an α -orientation of the acetoxy group at C-2 rather than the β -orientation in britanin. Consequently, the structure of **6** was determined to be 2 α ,6 α -diacetoxy-4 β -hydroxy-11(13)-pseudoguaiaen-12,8 α -olide.

By comparing physical and spectroscopic data with literature values, the eight known sesquiterpenes (**7–14**) were identified as britanin C (**7**),¹⁰ 4-epi-isoinuscolide (**8**),¹¹ gaillardin (**9**),⁶ xerantholide (**10**),¹² inuchinenolide A (**11**),¹³ tomentosin (**12**),⁶ 2-O-acetyl-4-epipulchellin (**13**),¹⁴ and 2-desoxy-4-epi-pucgellin (**14**).¹⁵

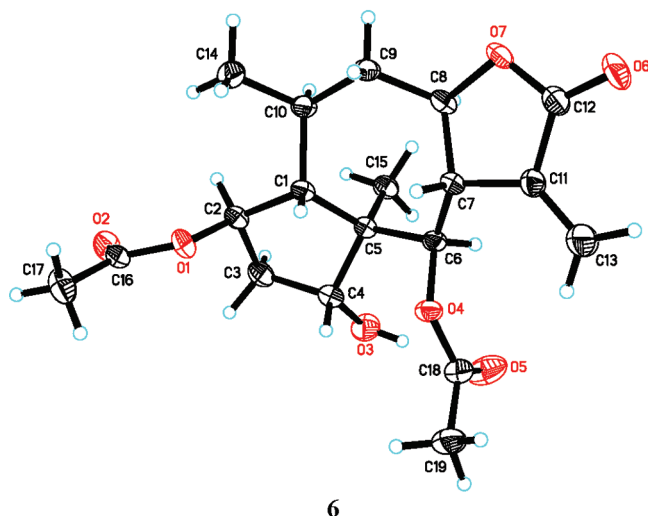
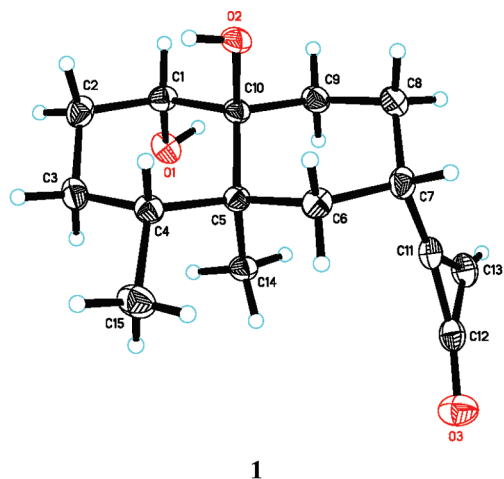


Figure 1. X-ray crystal structures of **1** and **6**.

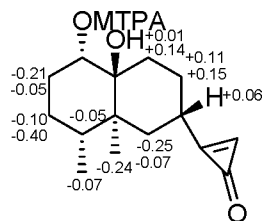


Figure 2. $\Delta\delta$ ($\delta_S - \delta_R$) values in ppm for MTPA derivatives of **1**.

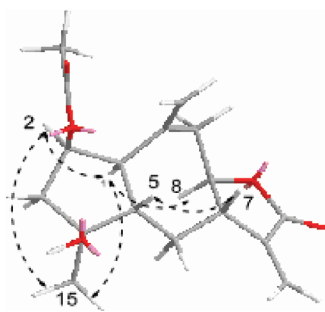


Figure 3. Selected NOESY correlations of **5**.

All 14 compounds were examined with respect to their inhibitory activities against LPS-induced NO production in RAW264.7 macrophages.^{16,17} The pseudoguaianolide-type sesquiterpenoids **6**, **13**, and **14** exhibited potent inhibitory activities against NO

Table 3. Inhibitory Effects of Sesquiterpenoids against LPS-Induced NO Production in RAW 264.7 Macrophages

compound	IC ₅₀ (μ M)
1	12.89
2	26.18
3	13.10
4	13.32
5	5.29
6	0.11
7	22.79
8	28.70
9	2.95
10	35.60
11	14.36
12	9.15
13	0.25
14	0.10
AG	2.70

^a Inhibitory effects against LPS-induced NO production in RAW 264.7 macrophages. AG: aminoguanidine.

production, with IC₅₀ values of 0.11, 0.25, and 0.10 μ M, respectively, compared to that of aminoguanidine, an iNOS inhibitor with an IC₅₀ value of 2.70 μ M (Table 3). Furthermore, these compounds were not significantly cytotoxic at the concentrations required for inhibiting NO production (as determined by MTT assay). Accordingly, these findings suggest that pseudoguaianolides should be viewed as promising lead compounds for the development of novel anti-inflammatory agents.

Experimental Section

General Experimental Procedures. Optical rotations were measured on a Perkin-Elmer-341 digital polarimeter (Perkin-Elmer, Norwalk, CT). IR spectra were recorded on a Bruker-Vector-22 spectrometer using KBr pellets. ESIMS was performed using an Agilent-1100-LC/MSD-Trap (ESIMS) and an Agilent Micro-Q-ToF, whereas HRESIMS was performed using a Waters Q-TOF micro mass spectrometer. Column chromatography (CC) was performed using silica gel (SiO₂, 200–300 mesh, Qingdao Haiyang Chemical & Special Silica Gel Co, Ltd., Qingdao, P. R. China) and Sephadex LH-20 (GE Healthcare Bio-Sciences AB, Sweden). TLC was performed using precoated SiO₂ GF₂₅₄ plates (Qingdao Haiyang Chemical & Special Silica Gel Co, Ltd., Qingdao, P. R. China), and compounds were visualized by exposure to UV light at 254 nm and spraying with a solution of vanillin in 5% H₂SO₄. Preparative HPLC using a Shimadzu PRC-ODS EV0233 and a C₁₈, 5 μ m, 300 \times 10 mm column and ¹H, ¹³C, and 2D NMR spectra were obtained using a Bruker-DRX-500 spectrometer.

Plant Material. The aerial parts of *I. linearifolia* were collected from Changfeng County, Anhui Province, People's Republic of China, in late July 2007, and authenticated by Prof. Shou-Jin Liu of Anhui University of Traditional Chinese Medicine. A herbarium specimen (No. XX20070701) was deposited at the School of Pharmacy, Shanghai Jiao Tong University.

Extraction and Isolation. The air-dried and powdered aerial parts of *I. linearifolia* (60.0 kg) were extracted with 95% EtOH three times each for 24 h at room temperature. The extract was concentrated, and the residue was suspended in H₂O and then partitioned successively with petroleum ether (PE), CHCl₃, EtOAc, and *n*-BuOH. A portion of the PE extract (116.8 g from 1000.0 g) was subjected to silica gel column chromatography (100 mesh, 1000 g) and eluted with a PE–EtOAc (100:1 \rightarrow 1:2) gradient to give 14 fractions (Fr.1–Fr.14) based on TLC analysis. Fr.5 and Fr.6 were chromatographed on silica gel using PE–EtOAc (20:1) as eluent to yield subfractions 5a–5e and 6a–6f, respectively. Subfractions 5e and 6c were further purified by preparative HPLC (MeOH–H₂O). Subfraction 5e gave **11** (*t*_R 32.3 min, 29.0 mg) (MeOH–H₂O, 40:60, flow rate 10 mL/min), whereas 6c yielded **3** (*t*_R 26.1 min, 2.0 mg), **4** (*t*_R 20.2 min, 4.8 mg), **5** (*t*_R 30.8 min, 4.8 mg), and **9** (*t*_R 35.2 min, 7.1 mg) (MeOH–H₂O, 35:65, flow rate 10 mL/min). Subfraction 6d was subjected to a silica gel column using PE–EtOAc (15:1) as eluent to yield **7** (6.6 mg). Compound **1** (31.0 mg) was crystallized (MeOH) from Fr.10, and **6** (330.2 mg) was crystallized (MeOH) from Fr.1. A portion of the CHCl₃ fraction (240.0 g from 600.0 g) was chromatographed over silica gel (100 mesh,

2000 g) and eluted with increasing amounts of MeOH (0–100%) in CH₂Cl₂ to afford 11 fractions (F1–F11). F1 and F2 were subjected to a Sephadex LH-20 column using MeOH as eluent to give subfractions 1A–1G and 2A–2J, respectively. Subfraction 1D was further separated into six fractions (1D1–1D6) using the same system. The subfractions 2C, 2D, 2E, 4D4, 1D4, and 1D6 were further separated by preparative HPLC (MeOH–H₂O). Subfractions 2C and 2E yielded **2** (*t*_R 25.6 min, 12.8 mg) and **13** (*t*_R 20.3 min, 10.2 mg) using MeOH–H₂O as eluent at 35:65 and 45:55 (flow rate 10 mL/min), respectively. Subfraction 2D yielded **12** (*t*_R 34.0 min, 38.6 mg) and **8** (*t*_R 36.2 min, 2.8 mg) (MeOH–H₂O, 40:60), and subfraction 1D6 yielded **10** (*t*_R 28.1 min, 7.3 mg) (MeOH–H₂O, 60:40). Subfraction 4D4 was further purified to give **14** (*t*_R 18.4 min, 4.0 mg) (MeOH–H₂O, 55:45).

Compound 1: monoclinic crystals (MeOH); [α]_D²⁰ –16.2 (*c* 1.2, MeOH); IR (KBr) ν_{\max} 3450, 3268, 1845, 1797, 1627, 1556, 1382, 1081, 944 cm⁻¹; ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS *m/z* 523.3 [2 M + Na]⁺, *m/z* 499.3 [2 M – H]⁻; HRESIMS (positive ion) *m/z* 251.1651 [M + H]⁺ (calcd for C₁₅H₂₃O₃, 251.1647).

Compound 2: light yellow gum; [α]_D²⁰ +46.7 (*c* 0.5, MeOH); IR (KBr) ν_{\max} 3427, 1738, 1456, 1100 cm⁻¹; ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS *m/z* 345.1 [M + Na]⁺, *m/z* 687.2 [2 M – H]⁻; HRESIMS (negative ion) *m/z* 321.1339 [M – H]⁻ (calcd for C₁₇H₂₁O₆, 321.1339).

Compound 3: colorless gum; [α]_D²⁰ –10.9 (*c* 0.3, MeOH); ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS *m/z* 287.2 [M + Na]⁺, *m/z* 263.1 [M – H]⁻; HRESIMS (positive ion) *m/z* 265.1445 [M + H]⁺ (calcd for C₁₅H₂₁O₄, 265.1440).

Compound 4: light yellow oil; [α]_D²⁰ –2.0 (*c* 0.5, MeOH); ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (positive ion) *m/z* 343.3 [M + Na]⁺; HRESIMS (positive ion) *m/z* 343.1542 [M + Na]⁺ (calcd for C₁₈H₂₄O₅Na, 343.1521).

Compound 5: light yellow oil; [α]_D²⁰ +27.9 (*c* 0.2, MeOH); ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (positive ion) *m/z* 329.2 [M + Na]⁺; HRESIMS (negative ion) *m/z* 305.1402 [M – H]⁻ (calcd for C₁₅H₂₁O₅, 305.1389).

Compound 6: colorless, tetragonal crystals (MeOH); [α]_D²⁰ –23.2 (*c* 1.0, CHCl₃); ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS *m/z* 389.2 [M + Na]⁺, *m/z* 365.0 [M – H]⁻; HRESIMS (positive ion) *m/z* 367.1767 [M + H]⁺ (calcd for C₁₉H₂₇O₇, 367.1757).

Crystallographic data of compound **1**: C₁₅H₂₃O₃, *M* = 250.33, monoclinic, space group *P*2(1), *a* = 7.6657(9) Å, α = 90°; *b* = 10.1466(12) Å, β = 111.211(2)°; *c* = 9.1327(11) Å, γ = 90°; *V* = 662.23(14) Å³, *Z* = 2, *D*_{calc} = 1.255 mg/m³, crystal size 0.425 × 0.330 × 0.278 mm³. Mo K α (λ = 0.71073 Å), *F*(000) = 272, *T* = 293(2) K. The final *R* values were *R*₁ = 0.0361 and *wR*₂ = 0.0884 for 3988 observed reflections [*I* > 2 σ (*I*)].

Crystallographic data of compound **6**: C₁₉H₂₆O₇, *M* = 366.43, tetragonal, space group *P*4(3)2(1)2, *a* = 9.3534(7) Å, α = 90°; *b* = 9.3534(7) Å, β = 90°; *c* = 42.209(5) Å, γ = 90°; *V* = 3692.7(6) Å³, *Z* = 8, *D*_{calc} = 1.318 mg/m³, crystal size 0.320 × 0.285 × 0.197 mm³. Mo K α (λ = 0.71073 Å), *F*(000) = 1568, *T* = 293(2) K. The final *R* values were *R*₁ = 0.0458 and *wR*₂ = 0.1013 for 22 084 observed reflections [*I* > 2 σ (*I*)].

Crystallographic data for **1** and **6** have been deposited in the Cambridge Crystallographic Data Centre (deposition number: CCDC 762277, 762278). Copies of the data may be obtained free of charge on application to the Director, CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: +44-(0)1223-336033 or e-mail: deposit@ccdc.cam.ac.uk).

Preparation of (R)- and (S)-MTPA Esters of 1. The (R)- and (S)-MTPA esters of **1** were prepared using a modified Mosher method.^{4,5} Briefly, **1** (3 mg) was treated with (R)- or (S)-MTPACl (each 5 μ L) in

pyridine-*d*₅ containing a small amount of DMAP in a clean NMR tube under an N₂ stream, and the NMR tube was immediately shaken to mix the sample and MTPACl. The tube was placed in a water bath at 50 °C for 4 h to afford the (S)-MTPA ester derivative (**1a**)/(R)-MTPA ester derivative (**1b**). The ¹H NMR data of the (S)-MTPA ester derivative (**1a**) and (R)-MTPA ester derivative (**1b**) were obtained directly from the reaction mixture (pyridine-*d*₅, 500 MHz).

Assay for Inhibitory Activities against LPS-Induced NO Production. This assay was conducted as previously described.^{16,17} Briefly, RAW264.7 cells grown on a 100 mm culture dish were harvested and seeded in 96-well plates at 2 × 10⁵ cells/well for NO production. The plates were pretreated with various concentrations of samples for 30 min and then incubated for 24 h with or without 1 μ g/mL of LPS. The nitrite concentration in the culture supernatant was measured by the Griess reaction. Cell viability was measured by a MTT [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide] assay (Sigma-Aldrich).

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Supporting Information Available: 1D and 2D NMR spectra of compounds **1** to **6** and the X-ray crystallographic data (CIF files) of **1** and **6** are available free of charge via the Internet at <http://pubs.acs.org>.

References and Notes

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